TREATING OMICRON BA.4 & BA.5 VIA HERBAL ANTIOXIDANT ASAFOETIDA: A DFT STUDY OF CARBON NANOCARRIER IN DRUG DELIVERY

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ABSTRACT

Currently, the results of researches have exhibited that Omicron sub-lineages BA.4 and BA.5, evaluated to BA.1 and BA.2, deserted neutralization from sera of triple vaccinated particulars to a bigger extension. Therefore, a novel wave of Omicron virus has appeared driven by BA.4 & BA.5 subvariants. Nanocarriers contain carbon atoms with functional nanostructures, which not only help the improved mechanical properties but also indicate the bioactivities for regulating cell status. In this research, asafoetida as a medicinal plant can be applied in treatment for Omicron subvariants BA.4 and BA.5 through adsorbing of its effective compound of ferulic acid on the surface of (6,6) armchair single-walled carbon nanotube as the drug delivery model due to direct electron transfer principle which has been studied by density functional theory (DFT) methods.

On the other hand, it has been accomplished the B3LYP/6-311+G (d,p) level of theory to evaluate the aptitude of SWCNT for adsorbing effective compound in asafoetida medicinal plant through nuclear magnetic resonance and thermodynamic parameters. In fact, the achieved results have represented that the feasibility of using (6,6) armchair SWCNT and ferulic acid becomes the norm in drug delivery system which has been attained by quantum calculations due to physico-chemical properties of NMR, IR and UV-VIS spectroscopy. Besides, the energy gap analysis of HOMO-LUMO has illustrated the charge distribution in the frontier molecular orbitals of ferulic acid in asafoetida drug through adsorption on the surface of (6,6) armchair carbon nanotube (CNT).

Keywords: Omicron subvariants BA.4 and BA.5, Asafoetida; Ferulic acid; CNT; drug delivery.

1. INTRODUCTION

During the COVID-19 pandemic, the new BA.2 subvariant has quickly substituted by old subvariants BA.1 and BA.1.1. The latter finding of three serious Omicron subvariants has increased in some concerns [1-7].

Besides, in the United States, the subvariant of BA.2.12.1 entered then was developed considerably in the northeast zone. Omicron subvariants of BA.4 and BA.5 appeared in South Africa became main variants. These new Omicron subvariants have been detected worldwide, though at low levels presently. However, their growth trajectories in the U.S. and South Africa exhibit an important transmission advantage.

The replacements in the BA.4 & BA.5 RBDs, L452R, and F486V are the most problem because of their possibility to allow immune attack. On the other hand, both these mutations are close to the angiotensin-converting enzyme 2 (ACE2) receptor surface, therefore might modify RBD-ACE2 association and the neutralizing capacity of natural and vaccine acquired immunity. While the reversion mutation Q493, which also exists in the ACE2, likely decreases the escape from respond to previous SARS-CoV-2 types. Currently, major variants of COVID-19 pandemic have been introduced as SARS-CoV-2 Omicron subvariants BA.4 and BA.5. The spike protein of BA.2.12.1 42 includes L452Q and S704L alterations plus to the recognized mutations in BA.2, while the 43 spike proteins of BA.4 and BA.5 are similar [8-11].

For decades, herbs and medicinal plants for decades have been considered as treatment viral diseases treatment. These medicinal plants could apply as strong origins for novel antiviral drugs versus rising and increasing viral diseases such as COVID-19 and different subvariants of Omicron virus. In this work, it has been studied that asafoetida might be an inhibitor of the Omicron subvariants BA.4 and BA.5 which is exhibited for the cure of sick people. The impact of asafoetida might decrease in patients having an impaired immune system. Asafoetida is a perennial herb with a little thick root, fleshy, cut and dusty leaves, fleshy stem, yellow flowers with umbrella-like groups and up to two meters height [12, 13].

The medicinal organ of this plant is a resinous gum obtained by cutting the upper parts of the root. Resin, essential oil, gum, ferulic acid and disulfide compounds are among the effective ingredients of asafoetida. The medicinal function of this plant is anti-flatulent and expectorant and it is used in the treatment of indigestion or colic related to flatulence, bronchitis, cough and nervous disorders [12]. The substantial amounts of native species of asafoetida are grown in the mountains of Afghanistan and the deserts of Iran [14].

Therefore, we have investigated the structural, electronic, physical and chemical characteristics of ferulic acid which is one the most effective compounds in asafoetida that might be applied to treat the Omicron subvariants BA.4 and BA.5 through relieving different side-effects in human body [Scheme1].

Scheme 1. Herbal antioxidant of asafoetida, ribbons PDB and Ramachandran plot as the inhibitor of SARS-CoV-2 Omicron BA.4 and BA.5 variant spikes for relieving different side-effects in human body.

There is an attention to enhancing the bioavailability and duration of action for a drug to modify therapeutic consequences. Drug delivery technique is able to change a drug’s pharmacokinetics and specificity by formulating it with various

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ingredients, drug carriers, and the medical equipment [15-19]. Nanomedicine in
drug delivery is for achieving the improved delivery of water insoluble drugs,
delivery of large macromolecule drugs to intracellular sites of action, and
codelivery of two or more drugs or therapeutic agents for combination remedy
[20-22].

Nanotubes with their intrinsic properties have been considered potential
candidates for drug delivery carriers. The capped ends of nanotubes may be
opened up by oxidation, allowing for the insertion of molecules of interest inside
the nanotube. Carbon nanotubes (CNTs) can easily penetrate cells,
delivering drugs directly to the cytoplasm or nucleus. Nanotubes conform to a
perpendicular position with the cell membrane during uptake, perforating and
diffusing through the lipid bilayer to enter the cytoplasm. Functionalized CNTs
are easily internalized by cells through passive and endocytosis-independent
mechanisms [23-30].

In this research, we have ferulic acid in asafoetida adsorbed on surface of (6,6)
armchair CNT in water medium for preventing the activity of Omicron
subvariants BA.4 and BA.5 (Scheme2).

![Scheme2. Mechanism of adsorption of ferulic acid on the surface of (6,6) armchair SWCNT in water medium.](image)

The structure of ferulic acid has been studied in this article which is an
effective component in the herbal antioxidant of asafoetida for adsorption on the
surface of (6,6) armchair CNT through drug delivery methodology (Scheme2).
Therefore, a DFT method has been run for finding the optimized coordination of
ferulic acid adsorbed on the surface of (6,6) armchair CNT by using Gaussian
16 revision C.01 program [31].

2. THEORETICAL BACKGROUND, MATERIAL AND METHOD

In this article, the geometry coordination of ferulic acid in asafoetida has been
optimized at the framework of DFT using the three-parameter Becke’s exchange
[32] and Lee-Yang-Parr’s correlation non-local functional [33], usually known
as B3LYP method and basis set of 6-311+G(d,p). The density functional theory
(DFT) is one of the most employed approximations of Hohenberg, Kohn and
Sham which allows the theoretical study of material properties [34]. Density
functional theory (DFT) represents an advantageous methodology for estimating
the chemical systems, and discovering its similarities and differences to other
computational employed methodologies [35, 36].

Therefore, it has been illustrated the electronic structure of adsorbed (6,6)
armchair CNT by ferulic acid in asafoetida medicinal plant for measuring
physico-chemical properties (Scheme2).

In this work, the Onsager model has been accomplished that was developed by
Frisch, Wong and Wiberg utilizes spherical cavities. Even though this implies a
less accurate description of the solute-solvent interface, this approximation
simplifies the evaluation of energy formatives in geometry optimizations, and
frequency analysis. Moreover, Cramer and Truhlar improved this model at the
dipole level [37-41]. In fact, a cavity must have a physical sense such as Onsager
model, and has a mathematical ability as often happened in other descriptions of
solvent impacts [42]. On the other hand, the cavity has to keep out the solvent
and including its frontiers as the biggest probability part of the solute charge
distribution [43-45].

Basically, a group of quantum theoretical methods has been run for exploring
some physical and chemical properties from optimized structure of ferulic acid
in asafoetida medicinal plant adsorbed on the surface of (6,6) armchair CNT
including charge distribution, thermodynamic calculations, nuclear magnetic
resonance analysis, frontier orbitals of HOMO & LUMO analysis and UV-VIS
discussion due to designing a drug delivery model y using Gaussian 16 revision
C.01 program [31].

Moreover, the gauge including atomic orbitals (GIAO) has been adopted to
solve the gauge problem in the calculation of nuclear magnetic shielding for
ferulic acid in asafoetida adsorbed on the surface of (6,6) armchair CNT using
density functional theory (DFT) calculation [46].

3. RESULTS AND DISCUSSION

Carbon nanotubes can easily penetrate cells, delivering drugs directly to the
cytoplasm or nucleus. These compounds describe drug delivery platforms that
may be functionalized with various biomolecules containing antibodies, proteins,
and DNA. This permits the particular target for transferring the special tissues,
organs, or cells. Drug delivery systems improve the pharmacological and
therapeutic profile and efficacy of the drug and lower the occurrence of off-
targets.
3.1. Analysis of NMR spectroscopy

Chemical shielding eigenvalues of $\sigma_{11}$, $\sigma_{22}$, $\sigma_{33}$ (ppm), isotropic shielding tensor ($\sigma_{iso}$), and anisotropic shielding tensor ($\sigma_{aniso}$) for ferulic acid in asafoetida medicinal plant adsorbed on the surface of (6,6) armchair SWCNT, respectively, have been obtained. The calculations have been accomplished based on B3LYP/6-311+G (d,p) level of theory by using Gaussian 16 revision C01 program (Table 1).

SCF GIAO magnetic shielding tensor in ppm for H, C, O atoms and exploring the active site of ferulic acid in asafoetida as the natural drug for treatment the Omicron subvariants BA.4 and BA.5 has been drawn (Figure 1a&b).

Table 1. Chemical shielding of eigenvalues $\sigma_{11}$, $\sigma_{22}$, $\sigma_{33}$ (ppm), isotropic shielding tensor ($\sigma_{iso}$) and anisotropic shielding tensor ($\sigma_{aniso}$), also atomic charge (Q) of H, C, O atoms in ferulic acid extracted from asafoetida medication using SCF GIAO theoretical method.

<table>
<thead>
<tr>
<th>Atom</th>
<th>$\sigma_{11}$</th>
<th>$\sigma_{22}$</th>
<th>$\sigma_{33}$</th>
<th>$\sigma_{iso}$</th>
<th>$\sigma_{aniso}$</th>
<th>Q</th>
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<tr>
<td>C1</td>
<td>-15.6372</td>
<td>161.9188</td>
<td>172.1630</td>
<td></td>
<td></td>
<td>99.0222</td>
</tr>
<tr>
<td>C2</td>
<td>44.5827</td>
<td>160.8814</td>
<td>220.0070</td>
<td>141.8237</td>
<td>117.2749</td>
<td>-0.0972</td>
</tr>
<tr>
<td>C3</td>
<td>18.7049</td>
<td>131.5717</td>
<td>212.1846</td>
<td>120.8204</td>
<td>137.0463</td>
<td>-0.0181</td>
</tr>
<tr>
<td>C4</td>
<td>78.6689</td>
<td>113.1457</td>
<td>242.8672</td>
<td>144.8939</td>
<td>146.9599</td>
<td>-0.0325</td>
</tr>
<tr>
<td>C5</td>
<td>68.8765</td>
<td>117.8510</td>
<td>231.2939</td>
<td>139.3404</td>
<td>137.9302</td>
<td>-0.0509</td>
</tr>
<tr>
<td>C6</td>
<td>83.5398</td>
<td>99.7001</td>
<td>197.7125</td>
<td>126.9841</td>
<td>106.0926</td>
<td>0.1129</td>
</tr>
<tr>
<td>C7</td>
<td>73.3400</td>
<td>94.6507</td>
<td>203.9832</td>
<td>123.9730</td>
<td>119.9329</td>
<td>0.1122</td>
</tr>
<tr>
<td>C8</td>
<td>63.7010</td>
<td>136.2266</td>
<td>218.9721</td>
<td>139.6332</td>
<td>119.0084</td>
<td>-0.0920</td>
</tr>
<tr>
<td>C9</td>
<td>55.0970</td>
<td>117.9162</td>
<td>222.1804</td>
<td>131.7312</td>
<td>135.6738</td>
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<tr>
<td>O10</td>
<td>267.8883</td>
<td>379.3201</td>
<td>404.3846</td>
<td>350.5310</td>
<td>80.7804</td>
<td>-0.2543</td>
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<tr>
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<td>178.7827</td>
<td>231.0085</td>
<td>193.3171</td>
<td>56.5370</td>
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<tr>
<td>O12</td>
<td>298.6211</td>
<td>317.3092</td>
<td>326.7566</td>
<td>314.2290</td>
<td>18.7914</td>
<td>-0.2976</td>
</tr>
<tr>
<td>O13</td>
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<td>-260.7962</td>
<td>442.1589</td>
<td>-166.6106</td>
<td>913.1543</td>
<td>-0.2909</td>
</tr>
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<td>O14</td>
<td>146.0424</td>
<td>301.2683</td>
<td>338.0813</td>
<td>261.7973</td>
<td>114.4259</td>
<td>-0.3178</td>
</tr>
<tr>
<td>H15</td>
<td>19.9683</td>
<td>25.2453</td>
<td>32.1288</td>
<td>25.7808</td>
<td>9.5220</td>
<td>0.0812</td>
</tr>
<tr>
<td>H16</td>
<td>19.6684</td>
<td>24.8934</td>
<td>31.4550</td>
<td>25.3390</td>
<td>9.1741</td>
<td>0.0793</td>
</tr>
<tr>
<td>H17</td>
<td>17.1551</td>
<td>26.7581</td>
<td>31.8257</td>
<td>25.2463</td>
<td>9.8691</td>
<td>0.0770</td>
</tr>
<tr>
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<td>21.8600</td>
<td>29.3262</td>
<td>29.8229</td>
<td>27.0300</td>
<td>4.2298</td>
<td>0.0540</td>
</tr>
<tr>
<td>H19</td>
<td>20.5091</td>
<td>28.2600</td>
<td>29.4811</td>
<td>26.0834</td>
<td>5.0966</td>
<td>0.0643</td>
</tr>
<tr>
<td>H20</td>
<td>24.6065</td>
<td>27.1074</td>
<td>37.9428</td>
<td>29.8856</td>
<td>12.0859</td>
<td>0.0706</td>
</tr>
<tr>
<td>H21</td>
<td>24.9868</td>
<td>29.2757</td>
<td>35.5131</td>
<td>29.9252</td>
<td>8.3819</td>
<td>0.0683</td>
</tr>
<tr>
<td>H22</td>
<td>25.0105</td>
<td>29.2715</td>
<td>35.5234</td>
<td>29.9351</td>
<td>8.3824</td>
<td>0.0681</td>
</tr>
<tr>
<td>H23</td>
<td>22.7081</td>
<td>29.1619</td>
<td>40.3955</td>
<td>30.7552</td>
<td>14.4605</td>
<td>0.2170</td>
</tr>
<tr>
<td>H24</td>
<td>22.9685</td>
<td>24.8511</td>
<td>35.9453</td>
<td>27.9216</td>
<td>12.0355</td>
<td>0.2208</td>
</tr>
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</table>

The ferulic acid has approximately shown the fluctuation behavior (30-350 ppm) for various atoms in the active sites of these compounds through the NMR properties (Figure 1a). The sharpest peak of NMR spectrum for ferulic acid has been almost observed in 30 ppm, and the weakest peaks of NMR spectrum have approximately appeared in about 100 ppm, 150 ppm, 200 ppm, 250 ppm, 300 ppm, and 350ppm, respectively (Figure 1a).

The ferulic acid adsorbed on the (6,6) armchair CNT has shown the chemical shielding including $\sigma_{11}$, $\sigma_{22}$, $\sigma_{33}$ and $\sigma_{iso}$, $\sigma_{aniso}$ (ppm) for various atoms of H, C, O in the active sites of the molecule through the NMR graph (Figure 1b). The most fluctuations of atomic charge and chemical shielding have been observed in O10, O12, O13, O14, respectively (Table 1), due to exploring the most electronegative atoms for adsorbed on the surface of (6,6) armchair CNT which represent the maximal shift in TMS B3LYP/6-311+G(d,p) (Figure 1a&b).

![Figure 1](image1.png)

Figure 1. a) NMR spectrum with SCF GIAO method. b) Chemical shielding of isotropic ($\sigma_{iso}$) and anisotropic ($\sigma_{aniso}$) calculated by level of theory B3LYP/6-311+G(d,p) for ferulic acid adsorbed on the surface of (6,6) armchair CNT.

In fact, the chemical shift tensors are achieved by the quantum chemical calculations in principal axes system to guess the isotropic chemical-shielding and anisotropic chemical-shielding [47-49].

3.2. Atomic Charge diffusion

The optimized atomic charge (Q) of H, C, O atoms (Table 1) in a polar medium of water solution indicates the stability of ferulic acid in asafoetida drug joint to the surface of (6, 6) armchair CNT as a drug delivery method for healing the Omicron subvariants BA.4 and BA.5 (Figure 2). Besides, it has been exhibited the Electrostatic Potential Map (ESP) which indicates the region including attractive - repulsive force of a fixed charged at different points in space that are parallel from a molecular surface of ferulic acid (Figure 2).

![Figure 2](image2.png)

Figure 2. The atomic charge (Q) of H, C, O atoms in the active sites of ferulic acid in asafoetida medicinal plant accompanying ESP map.

It has been recommended the reason for existing stability of ferulic acid adsorbed on the surface of (6,6) armchair SWCNT which is principally bound to the position of active sites of labeled H,C,O which move the charge of electrons in this compound in polar water molecules (Figure 2).
On the other hand, the electrophilic side chains of ferulic acid in asafoetida medicinal plant conduct us to find the reason for the activity and the stability of this drug against the Omicron subvariants BA.4 and BA.5 in human body.

3.3. Analysis of Infrared spectrum

It has been resulted the infrared (IR) spectrum for ferulic acid adsorbed on the surface of (6,6) armchair CNT using B3LYP method and 6-311+G(d,p) basis set for atoms including H, C, O atoms for obtaining the more accurate equilibrium geometrical parameters, thermodynamic properties. The IR spectrum for ferulic acid in asafoetida drug has been shown in the frequency range about 50 cm⁻¹-4500 cm⁻¹ (Figure 3). It can be seen that the strongest allowed peaks with highest intensities occur about 50 cm⁻¹, 150 cm⁻¹,250 cm⁻¹,625 cm⁻¹, 1400 cm⁻¹, 1450 cm⁻¹, 1550 cm⁻¹,1650 cm⁻¹,1700 cm⁻¹, 1750 cm⁻¹, 1800 cm⁻¹, 2000 cm⁻¹, 2100 cm⁻¹, 2150 cm⁻¹, 2200 cm⁻¹, and 3500 cm⁻¹ respectively (Figure3).

**Figure3.** The infrared spectrum of ferulic acid as the effective compound in asafoetida drug adsorbed on the surface of (6,6) armchair CNT using 6-311+G(d,p) calculations.

Figure3 demonstrates the reason for existing observed various results frequencies of ferulic acid adsorbed on the surface of (6,6) armchair CNT which presents the position of active sites of labeled H, C, O atoms in asafoetida medication. The calculations of the relative harmonic frequencies, IR intensities in various normal modes and thermodynamic properties of ∆H, ∆G, ∆S for ferulic acid adsorbed on the surface of (6,6) armchair CNT using B3LYP/6-311+G(d,p) method have been reported in Table2.

Table 2. The results of calculated functions of harmonic frequencies (cm⁻¹), IR intensities (km/mol) in different normal modes; thermodynamic properties of ∆G, ∆H in kcal/mol and ∆S in cal/mol.K⁻¹ at 300K for ferulic acid.

<table>
<thead>
<tr>
<th>Normal mode</th>
<th>IR Intensity</th>
<th>Frequency</th>
<th>∆G ×10⁻³</th>
<th>∆H×10⁻³</th>
<th>∆S</th>
<th>Dipole moment</th>
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<td>7</td>
<td>20.1093</td>
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<td>-2.3487</td>
<td>93.49</td>
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</table>

The polarization functions into the applied basis set in the computations always demonstrate a significant achievement on the simulation and modeling methods of theoretical levels. The normal modes of IR spectrum have been exploring the harmonic potential wells by analytic methods which keep the movement of all atoms at the same time in the vibration time scale leading to a natural definition of molecular vibrations (Table2).

The results of the above observations strongly suggest that ferulic acid in asafoetida medicinal plant adsorbed on the surface of (6,6) armchair SWCNT at B3LYP/6-311+G(d,p) method in water solvent is predominantly due to basis set functions which are induced by a change in polarity of the environment which has approved that an increase in the dielectric constant increases the stability and efficiency of this drug for treating Omicron subvariants BA.4 and BA.5.

3.4. Frontier orbitals of HOMO & LUMO

The highest occupied molecular orbital energy (HOMO) and the lowest unoccupied molecular orbital energy (LUMO) have been calculated for ferulic acid (Table3). The HOMO, LUMO and band energy gap (ev) indicated the pictorial explanation of the frontier molecular orbital’s and their respective positive and negative zones which are an important factor for identifying the molecular characteristics of ferulic acid in asafoetida medicinal plant.

Table3. The HOMO (a.u.), LUMO (a.u.), band energy gap (ev) and other quantities (ev) for ferulic acid in asafoetida drug.

<table>
<thead>
<tr>
<th>Compound</th>
<th>(eV)</th>
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</thead>
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<td>µ</td>
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</tr>
<tr>
<td>Z</td>
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<tr>
<td>η</td>
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</tr>
<tr>
<td>ζ</td>
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<tr>
<td>ζ/η</td>
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</tbody>
</table>

In other words, the HOMO shows the capability for giving an electron while the LUMO as an electron acceptor exhibits the capability for achieving an electron. Therefore, the energy gap (∆E = E LUMO - E HOMO) indicates the energy difference between frontier HOMO and LUMO orbital introducing the stability for the structure and unravels the chemical activity of the molecule. In this work, energy gap establishes how ferulic acid interacts with the surface of (6, 6) armchair CNT. Besides, frontier molecular orbitals run an important function in the optical and electrical properties like in UV-Vis spectra [50]. Moreover, for getting more conclusive approving in identifying the compound characteristics of this structure, a series of chemical reactivity parameters consisting of chemical potential (µ), electronegativity (η), hardness (ζ), softness (ζ), electrophilicity index (ω) has been done (Table3) [51-53].

The negative amounts of the chemical potential (µ) and the positive values of other quantities have exhibited a good stability of ferulic acid in asafoetida drug through adsorption on the surface of (6,6) armchair CNT correlated with the trend of drug delivery (Table3). This stable complex and its binding pose within Omicron subvariants BA.4 and BA.5 viruses as an illustration of the molecular drug delivery.

3.5. Analysis of UV-VIS spectroscopy

There is a critical factor as an energy gap between HOMO and LUMO for recognizing the characteristics of molecular electrical transport [54-58]. Based on Frank–Condon principle, the maximum absorption peak (max) is related to an UV–visible spectrum to vertical excitation.

In this research, TD-DFT/6-311+G (d,p) calculations have been done to identify the low-lying excited states of ferulic acid in asafoetida medicinal plant. The results contain the vertical excitation energies, oscillator strength and wavenumber which have been presented in Figure4.
In the calculated value of UV-VIS spectrum for ferulic acid extracted from asafoetida drug, there are two maximum adsorption bands between 200 nm–400 nm. The strong adsorptions have been observed in 245 nm and 310 nm, respectively (Figure 4).

CONCLUSION

Omicron subvariants BA.4 and BA.5 which are stimulating a new wave of infections in South Africa, might finally erupt in all over the world. BA.4 & BA.5 subvariants become developed from BA.2 and have identical sequences of SARS-CoV-2 spike (S) glycoprotein.

In this article, treating the Omicron subvariants BA.4 and BA.5 has been studied through asafoetida medicinal plant due to ferulic acid extracted this natural drug. Ferulic acid has been calculated through adsorbing on the surface of (6,6) armchair SWCNT at B3LYP/6-311+G (d,p) level of theory in water medium as the drug delivery method.

Asafoetida has attracted much attention for the clinical treatment of Omicron subvariants BA.4 and BA.5 through adsorption of its effective compound of ferulic acid on the surface of (6,6) armchair SWCNT which introduces an efficient drug delivery system though charge distribution, NMR and IR spectroscopy on the optimized structure.

Moreover, the lowering of the energy gap (∆E = E_{LUMO} - E_{HOMO}) has illustrated the charge transfer interactions taking place within ferulic acid. The atomic charges have donated the proper perception of the molecular theory and the energies of fundamental molecular orbitals.

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